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### Chemistry Matter and Change Chapter 14: Mixtures and Solutions

Solutions Manual Chemistry: Matter and Change • Chapter 9 145 CHAPTER 9 SOLUTIONS MANUAL 35. Aqueous solutions of potassium iodide and silver nitrate are mixed, forming the precipitate silver iodide. chemical equation:  $KI(aq) + AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s)$  complete ionic equation:  $K^+(aq) + I^-(aq) + Ag^+(aq) + NO_3^-(aq) \rightarrow K^+(aq) + NO_3^-(aq) + AgI(s)$

### Chemical Reactions

Solution. Physical: boiling and melting are physical changes. When water boils, no bonds are broken or formed. The change could be written:  $\text{H}_2\text{O} (l) \rightarrow \text{H}_2\text{O} (g)$  Chemical: The dark grey nail changes color to form an orange flaky substance (the rust); this must be a chemical change.

### 3.6: Changes in Matter: Physical and Chemical Changes ...

240 Chemistry: Matter and Change • Chapter 12 Solutions Manual CHAPTER 12 SOLUTIONS MANUAL 30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

### States of Matter

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